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comparison with that of the oxygen of the ylide B.

Me R
$$\frac{h\nu}{N}$$
 $\frac{h\nu}{Cl}$ $\frac{h\nu}{Cl}$ $\frac{R=NH-NH}{2}$ $\frac{2a_{1}C}{Cl}$ $\frac{R=NH-NHMe}{Cl}$ $\frac{3a}{Cl}$

The facile introduction of nucleophiles in the 4 position ^{2b} and the possibilities of photorearrangement, render the isoxazolo [4,5-c] pyridine of valuable interest in the synthetic approach to a number of pyridine condensed heterocycles.

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References and Notes

- G.Adembri, A.Camparini, F.Ponticelli and P.Tedeschi, J.C.S. Perkin I, 1977,
 b) G.Adembri, A.Camparini, F.Ponticelli and P.Tedeschi, <u>ibid.</u>, in press;
 c) G.Adembri, A.Camparini, D.Donati, F.Ponticelli and P.Tedeschi, <u>Tetrahedron Lett.</u>, 1978, 4439.
- 2a) G.Adembri, A.Camparini, F.Ponticelli and P.Tedeschi, <u>J.C.S. Perkin I</u>, 1975, 2190; b) G.Adembri, A.Camparini, F.Ponticelli and P.Tedeschi, <u>J.Heterocycl.</u> Chem., 1979, 16, 49.
- 3) Anhydrous ether (<u>1a,c</u>) or ethanol (<u>1b,d</u>) solutions were irradiated with an unfiltered low pressure mercury lamp (<u>1a,c</u>) or a medium pressure mercury lamp, filtered with a CuSO₄ solution (<u>1b,d</u>). N₂ was bubbled through the solutions.

 In the case of <u>1c</u>, solution was continuously circulated through a silica gel column to stop the photoproduct <u>3a</u> in the column.
- 4) J.Fraser and E.Tittensor, <u>J.C.S.</u>, 1956, 1781.

between 4a and 4b added further support.

1-Aminopyrazolo [3,4-b] pyridine $\underline{5}$ was recovered as main product of irradiation of $\underline{1d}$. The structure of compound $\underline{5}$ was proved by transformation into 3-methyl pyrazolo [3,4-b] pyridine, prepared in an unambiguous way 7 .

Some differences in respect to the photobehaviour of isoxazol-5-ylhydrazines^{1c} are to be pointed out, mainly: i)isoxazol-5-ylhydrazines rearrange to 1-aminopyrazoles only by thermal^{1a}, but not by photochemical activation^{1c}; ii)no product deriving from 3-methyl shift through ketenimine intermediate was found after
irradiation of isoxazolopyridine <u>1a-d</u>.

Even if additional work is necessary in order to evidentiate the relative importance of different parameters, the results obtained can be rationalized in the scheme reported.

Initial N=0 bond opening leads to the intermediate A, which is trapped to 1-aminopyrazolo [3,4-b] pyridine $\underline{5}$ by unsubstituted α -hydrazinic nitrogen present at the 4 position of the isoxazolopyridine $\underline{1d}$. Otherwise rearrangement of intermediate A to the ylide B takes place. Pyrido=1,2,4-triazine $\underline{3a}$ or oxazolopyridines $\underline{2a}$,c are formed depending on the nucleophilic strength of 2-substituent in

comparison with that of the oxygen of the ylide B.

The facile introduction of nucleophiles in the 4 position ^{2b} and the possibilities of photorearrangement, render the isoxazolo [4,5-c] pyridine of valuable interest in the synthetic approach to a number of pyridine condensed heterocycles.

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- 3) Anhydrous ether (<u>1a,c</u>) or ethanol (<u>1b,d</u>) solutions were irradiated with an unfiltered low pressure mercury lamp (<u>1a,c</u>) or a medium pressure mercury lamp, filtered with a CuSO₄ solution (<u>1b,d</u>). N₂ was bubbled through the solutions. In the case of <u>1c</u>, solution was continuously circulated through a silica gel column to stop the photoproduct <u>3a</u> in the column.
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- 6) Preparation of compounds 1b,c was performed according to the procedure described in the ref. 2a for compound 1d.
- Compound 2a: m.p. 136-138° (sublimation in vacuo), δ (CDCl₃) 2.70(s,3,Me), 7.44 (s,1, H_7); compound 2c: m.p. 127-128° (ether/light petroleum), γ (KBr) 3280 cm⁻¹; δ (CDC1₃) 2.55(s,3,Me), 3.11(d, J=5.5, s with D₂0, 3, NMe), 5.27(exch. br. s,1,NH), 6.73(s,1,H₇); compound 2d: m.p. 80-81° (sublimation in vacuo), ν (KBr) 3240 cm⁻¹, δ (CDCl₃) 2.52(s,3,Me), 3.10(d,J=5.5,s with D₂0,3,NMe), 5.33 (exch. br. s,1,NH), 6.70(d,J=6.0,1,H₇), 7.98(d,J=6.0,1,H₆); $\underline{\text{compound } 3a}$: m.p. 185-188° (sublimation in vacuo), ν (KBr) 3100-2200 cm⁻¹, δ (CDCl₃) 2.27(s,3,Me), 2.87(s,3,NMe), 3.27(s,3,NMe), 6.46(s,1,H₆), 8.75(exch. br. s,1,NH/OH); compound 3b: δ (CDCl₃) 2.16(s,3,Me), 2.85(s,3,NMe), 3.20(s,3,NMe), 3.91(s,3,OMe), 6.58(s,1,H₆); compound 3c: m.p. 65-66° (sublimation in vacuo), δ (CDCl₃) 2.10 (s,3,Me), 2.86(s,3,NMe), 3.20(s,3,NMe), $6.92(dd,J=8.0 and <math>5.0,1,H_6)$, 7.23(dd,J=8.0)J=8.0 and $1.8,1,H_5$), 7.98(dd,J=5.0 and $1.8,1,H_7$); compound $4a:\nu$ (KBr) 3490, 3400 and 1660 cm⁻¹, δ (CDC1₃) 2.12 and 2.33(s,3,MeCO), 2.92(s,3,NMe), 3.02 and 3.08(s,3,NMe), 3.60(exch. br. s,2,NH₂), 3.89(s,3,0Me), 6.58(s,1,H₅); compound 4b: m.p. 118-120°, ν (KBr) 3460,3340,1640 cm⁻¹, δ (CDCl₃) 2.12 and 2.34(s,3, MeCO), 2.89(s,3,NMe), 3.03 and 3.09(s,3,NMe), 3.70(exch. br. s,2,NH₂), <math>6.80- $7.08(m, 2, H_4 \text{ and } H_5), 7.73-7.91(m, 1, H_6); \underline{\text{compound 5}}; m.p. > 350° (ethanol/$ water), ν (KBr) 3320,3195 and 3160-2200 cm⁻¹, δ (DMS-d₆) 2.46(s,3,Me), 6.10(exch. br. s,2,NH₂), 6.42(s,1,H₅), 11.77(exch. br. s,1,NH/OH). Satisfactory elemental analysis were obtained for all new compounds. Synthe-

tic details will be given in our full manuscript.

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